Quantitative description of the morphology of polyurethane nanocomposites for medical applications

J. Ryszkowska & B. Waśniewski

Warsaw University of Technology,
Faculty of Materials Science and Engineering, Poland

Abstract

This paper presents the application of stereology methods to the description of morphological properties of polyurethane nanocomposites for medical applications. The study of the cross-section surface structure of the obtained materials was performed by Atomic Force Microscopy. The volume of hard phase agglomerate was used to evaluate the degree of phase separation of the examined nanocomposites. The relationships between the domain agglomerate characteristics and the properties of nanocomposites obtained from them were analysed.

The results showed that nanocomposites with non-modified nanosilica dioxide (SiO$_2$) and nanosilica dioxide modified with NH$_2$ groups differs from polyurethane within the following properties: size and volume of the agglomerates of the hard domains, biocompatibility, thermo-mechanical and abrasive wear resistance.

Keywords: nanocomposites, polyurethane, structure, image analysis, biomedical application

1 Introduction

Polyurethanes (PURs) and their nanocomposites are a versatile plastic material, formulated to provide good biocompatibility, flexural endurance, high strength, high abrasion resistance and processing versatility over a wide range of applications [1]. The most common use in medical devices is in short-term implants [2]. Polyurethane elastomers are linear segmented copolymers
consisting of a relatively flexible component derived from a macrodiol called soft segment, and a relatively hard and stiff component derived from a diisocyanate and a chain extender called hard segment (Fig 1). Thermodynamic incompatibility of these segments leads to microphase separation. The domain structure formed by microscopic phase separation presents similar elastomeric properties to those shown for cross-linked rubber networks. The mechanical strength of this structure can be attributed to hard microdomains physically cross-linked through hydrogen bonding and dispersion forces, acting as filler-like reinforcement for the soft segment [3]. Polyurethanes are characterized by a complex morphology which is dependent upon the precise nature of the hard and soft segments and their composition, use of nanoparticles, preparation method and its parameters. All these factors influence the morphological factors such as degree of microphase separation, crystallinity, the domain agglomerate characteristics, and define properties such as hardness, stiffness, tensile strength, clarity and biocompatibility [1–7].

![Diagram of polyurethane structure](image)

**Figure 1:Domains structure of polyurethane.**

This paper presents the application of stereology methods to the description of morphological properties of polyurethane nanocomposites for medical application. Stereological parameters [8–10] chosen for analysis were used to evaluate the degree of phase separation of the examined nanocomposites. Relationships between the domain agglomerate characteristics and the properties of the nanocomposites obtained from them were analysed.
2 Experimental

2.1 Materials

The following reactants were used in the synthesis of PURs: polyoxythetramethylene glycol (PTMG) of molecular weight 2023 g/mol (Therathane® 2000) was supplied by Du Pont Nemours Co.; 4,4’-diphenylmethane diisocyanate (MDI) purchased from Aldrich Chemical Co. (Germany); ethylene glycol (GE) and glycerine (G) POCH (Gliwice, Poland). Polyol was dehydrated by mixing under vacuum for 2h at 120°C. Ethylene glycol and glycerin was dried under a molecular sieve.

Nanosilica powder and nanosilica powder modified with NH2 groups, with primary particle size of 75 nm were used as nanofiller (ICHP Warsaw). To prepare the composites, nanosilica – polyetherodiol 20% wt. concentrate was made. Firstly, polyetherodiol was melted down in the oven under 80°C degree. Then nanosilica was put into it and mixed with ultrasonic homogenizer VCX 750 by Sonics during 30min in pulse mode 3/3 (3sec mixing, 3sec stop).

2.2 Polyurethane and nanocomposites synthesis

Segmented polyether based polyurethanes with substrates molar ratio PTMG: MDI: GE: G equal to 40:80:27:24 (1:2:0.679:0.151), constant isocyanates index of 1.05 and with hard segment contents 20 wt.% was synthesized using a one-step polymerisation method. The polypol was cooled to 70°C±3°C, with glycol added and glycerin blended for 5min. Then the mixture was cooled to 60°C±3°C and MDI added. The samples were obtained with free casting method. The heat up process was carried out in the oven for 16 hours. To prepare the composites, appropriate quantities of concentrate were added to the polypol. The mixture was dehydrated in temperature of 120°C±5°C under 2-5 hPa pressure. Next the process was run in the same way as polyurethane synthesis. Description of achieved materials is presented in Table 1.

<table>
<thead>
<tr>
<th>Sample</th>
<th>0</th>
<th>1</th>
<th>2</th>
<th>3</th>
<th>1A</th>
<th>2A</th>
<th>3A</th>
</tr>
</thead>
<tbody>
<tr>
<td>Amount of nanofiller [wt.%]</td>
<td>0</td>
<td>0.5</td>
<td>1.0</td>
<td>2.0</td>
<td>0.5</td>
<td>1.0</td>
<td>2.0</td>
</tr>
<tr>
<td>Modification of nanofiller</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-NH2</td>
<td>-NH2</td>
<td>-NH2</td>
</tr>
</tbody>
</table>

2.3 Characterization

2.3.1 Dynamic mechanical analysis

Dynamic tests, by dynamic mechanical analysis (DMA) on a Thermal Instruments dynamic mechanical analyzer (Q800 TA), were also carried out in three-point bending mode on specimens with dimensions of 12 x 2 x 60mm³. Tests were performed with the amplitude of deformation during the bending of 25 µm. The frequency-dependent storage modulus was also evaluated with a
4.5–9 Hz frequency sweep at a constant temperature of 37°C with 1.5% strain and 0.01 N static force. Similar parameters were used by Hafeman and co-workers [11]. The storage modulus (E’) value was recorded as a function of frequency.

2.3.2 Abrasion resistance
The abrasion resistance of test samples was measured with a Schopper–Schlobach instrument with an APGi circulating roller from Heckert, and the procedure complied with the standard PN-ISO [12]. The test pieces are in the form of a roll (16 ± 0.2 in diameter and 2 mm high). Standard rubber from (The Institute for Engineering of Polymer Materials and Dyes, Elastomer and Rubber Technology Division in Piastów) was used as reference material [13]. The abrasion resistance index (ΔV) was calculated with the following relation:

\[ \Delta V = \frac{V_s}{V_t}, 100\% \]

where \( V_s \) is the loss of volume of the standard rubber (mm\(^3\)) and \( V_t \) is the loss of volume of the test sample (mm\(^3\)).

The density figures for the test pieces, which were necessary for calculations, were found by the method described in the standard PN-ISO [14].

2.3.3 Atomic Force Microscopy (AFM)
AFM images were recorded at 37°C in air using a Digital Instrument Multimode Nanoscope V (Digital Instruments, Santa Barbara, CA) operating in the tapping regime mode using antimony doped silicon cantilever tips (POL-15, 130 do 250 kHz, 48 N/m). Scanner was used with scan rates between 0.5 and 1 Hz. All images are subjected to a first-order plane-fitting procedure to compensate for sample tilt. The microstructure of polyurethane was investigated on micro sections. These were prepared using a rotary microtomy RM 2165 (Leica) with a LN 21 cooling device working at -60°C.

2.3.4 Image analysis
Binary images revealing hard domains agglomerates were produced via digital processing of AFM images. The size and volume fractions of the hard domains agglomerates in polyurethane and nanocomposites were determined by measurements on their sections [15]. Linear covariance method was used for the description of the distribution of particles. The image was transferred to MicroMeter software and quantitative analysis was performed. The diameters were randomly determined on 5 microphotographs.

2.3.5 Fourier transform infrared (FTIR) spectroscopy
Infrared spectra of PURs were collected using a FTIR spectrophotometer (Thermo Electron Corporation model Nicolet 6700). Measurements were carried out using attenuated total reflectance (ATR) technique. Each sample was scanned 64 times at a resolution of 4 cm\(^{-1}\) over the frequency range of 4000–400 cm\(^{-1}\). Analysis of FTIR data enabled to determine the carbonyl hydrogen-bonding index (R). A straight baseline was drawn in the spectrum between 1780
cm$^{-1}$ and 1640 cm$^{-1}$ and the carbonyl stretching zone was corrected by subtracting the baseline. To estimate the signal strengths, peak modeling of the infrared active carbonyl bands was carried out using the Gaussian curve-fitting method software OMNIC 7.3. The carbonyl absorption bands were deconvoluted into component bands, the peak area of these bands was measured and carbonyl hydrogen-bonding index $R$ was calculated using Eq. (1) [16, 17]:

$$R = \frac{A_{\text{bonded}}}{A_{\text{free}}}$$

Moreover, the degree of phase separation (DPS) was obtained through Eq. (2):

$$DPS = \frac{R}{R + 1}$$

3 Results and discussion

In polyurethanes application as short-term implants the modulus of elasticity and surface properties must very often be changed. The one way to solve the problem is the use of nanofiller [18–20]. Nanocomposites exhibit advantageous mechanical and physical properties already at small addition of modifying particles, frequently lower than 5 wt%. Most of nanofillers occur without any surface modification. However, sometimes we need to modify nanoparticles to obtain better dispersion of nanofiller in polymer matrix, for instance. One of the modifications is a chemical one relying on attaching functional groups i.e. –COOH, -NH$_2$, -NCO, -OH to nanofiller [20–24].

In this study were used polyetherourethanes with 0.5–2 wt% of nanosilica and nanosilica modified with amino groups.

The change in elasticity module was examined in the course of three-point bending as well as abrasion wear of fabricated materials (Table 2). Nanofiller introduction change the storage modulus and abrasion wear of polyurethane matrix.

| Table 2: Results of dynamic three point bending and abrasion wear of the PU and PU/SiO$_2$ composites. |
| Sample | 0 | 1 | 2 | 3 | 1A | 2A | 3A |
| E’, storage modulus by frequency 5Hz at 37°C, MPa | 24.6 | 2.8 | 32.6 | 30.8 | 3.4 | 40.6 | 41.6 |
| ΔV, abrasion wear, dm$^3$ | 35.2 | 26.0 | 27.7 | 48.2 | 23.0 | 27.3 | 31.8 |

The introduction of 0.5 wt% of nanofiller brings about a decrease in storage modulus and abrasion wear; bigger amounts of the filler make the two values grow. The use of modified nanofiller results in the fact that the obtained nanocomposites have a higher storage modulus at lower abrasion wear.

In order to explain the mechanism of the influence of nanofiller on polyurethane matrix structure microscopic observations were carried out of the cross-section surface of fabricated materials using AFM. An AFM tapping mode
is used to depict topographic features and the spatial variation in surface by height and phase imaging. For the examined materials the appearances of topographic and phase images are similar and show less obviously surface morphology related to hard domain and soft domain separation. An exemplary phase image of surface cross-section of examined PU is shown in Fig. 2a. Binary images of all face images of the polyurethanes and nanocomposites were performed and those selected from them were shown in Fig. 2 (b-f). The neat PU cut surface showed microphase separation in phase images, the hard phase of the polyurethane appears brighter than the soft phase. The hard phase during phase separation formed island domain like phase. The results in all nanocomposites are similar, showing a distinct topology of phase separation of hard and soft segments (Fig 2b-f). Images of Pu phase structure and nanocomposites are similar to the images obtained by Zhang et al. [21].

The AFM images could be explained as follows: Glycerine reacted with MDI and then formed island domain like hard phase. This usually accompanied chemical reactions such as cross-link, so the segments shrunk as a result [22] and formed caves like those in PU and nanocomposites.

The introduction of 0.5 wt% unmodified filler (Fig. 2c) makes the appearance of smaller islands, formed by hard phase, than in PU (Fig. 2b), visible in the pictures of cross-section surfaces of the nanocomposites. The use of 0.5 wt.% SiO$_2$ modified by NH$_2$ groups results in the growth of the size of islands formed by hard phase; moreover, they tend to merge into bigger agglomerates (Fig. 2d).

The size of the agglomerates of the hard domains was increased respectively to the nanofiller contents as shown in Fig. 2e and f c.

Nanda and his co-workers have reported small amounts of POSS usually tending to combine with hard segments and affect the properties of the hard segments of PU [23]. Changes in the morphology of polyurethanes, caused by the introduction of POSS, were also observed by Zhang and co workers [24]. It can be assumed that the mechanisms of changes caused by the use of nanosilica are similar. Quantitative analysis of binary pictures was used for the description of morphology of PU hard phase and its nanocomposites. Since hard phase created islands, often merging into bigger agglomerates, the agglomerates’ volume formed by hard phase was calculated and the results are presented in Fig. 3.

The difference in the level of interaction by hydrogen bond between the hard segments of the materials can regarded as an explanation of the differences in the volume of the agglomerates created by hard phase of the examined materials; hence the determination of hydrogen bond index (R) and phase separation index (DPS). The said difference in the level of interaction was determined basing on the vibrations of carbonyl groups occurring within the scope of wavenumbers 1760-1680cm$^{-1}$. The comparison of spectra in this scope for materials with equal nanosilica content modified with NH$_2$ groups is shown in Fig. 4.
Figure 2: AFM phase images at 5 μm scan sizes of (a) sample 0 and binary images: (b) sample 0 (c) sample 1 (d) sample 1A (e) sample 3 (f) sample 3A. Z ranges: 30°.

The R and DPS values of the polyurethane/nanosilica nanocomposites mixtures are given in Table 3. After the introduction of 0.5 wt.% of nanosilica, the R and DPS is higher than for polyurethane but modification causes a decrease in R and DPS. When a bigger amount of nanosilica is applied then R and DPS increase by adding –NH₂ group on the surface of nanosilica.

Several factors influence the DPS of polyurethane such as molecular weight, segmental length, crystallizability of soft segments, overall composition and intra- and inter-segments interactions [25–27]. The addition 0.5 wt% nanosilica causes weakening of the hydrogen-bond interactions created between polyurethane hard segments; segmental incompatibility in the polyurethane decreases by adding nanosilicas (Table 3). This explains the formation of the structure visible in the cross-section of the materials (Fig 2c, d). After the introduction of a bigger amount of the filler we observe no weakening of hydrogen-bond interactions created between polyurethane hard segments. A bigger amount of the filler increases incompatibility in the polyurethanes, causing a higher degree of phase separation in the polyurethane (Table 3).
Figure 3: Influence of nanosilica content on volume fraction of hard domains agglomerates in PUR: (1) nanosilica, (2) modified nanosilica.

Figure 4: Influence of modified nanosilica content on FTIR spectrum in the range 1760–1680 cm\(^{-1}\).

Table 3: Parameters of polyurethane phase separations.

<table>
<thead>
<tr>
<th>Sample</th>
<th>0</th>
<th>1</th>
<th>2</th>
<th>3</th>
<th>1A</th>
<th>2A</th>
<th>3A</th>
</tr>
</thead>
<tbody>
<tr>
<td>R</td>
<td>0.079</td>
<td>0.041</td>
<td>0.069</td>
<td>0.090</td>
<td>0.071</td>
<td>0.105</td>
<td>0.109</td>
</tr>
<tr>
<td>DPS</td>
<td>0.073</td>
<td>0.040</td>
<td>0.064</td>
<td>0.082</td>
<td>0.066</td>
<td>0.095</td>
<td>0.099</td>
</tr>
</tbody>
</table>

Moreover, in surface cross-sections of the examined nanocomposites bigger agglomerates of hard domains can be seen (Fig. 2e and f). Most probably, the increase in the degree of phase separation of this nanocomposites group can be attributed to the growth of the chain mobility in the polyurethane allowing the creation of more ordered phases, with respect to the polyurethane without nanosilica. A smaller amount of nanosilica weakened the interactions between hard segments but caused no growth of the chain mobility; it was only after adding bigger amounts of nanofiller (above 1%) that an increase in nanocomposites chain mobility was observed. Surface modification of nanosilica by NH\(_2\) groups results in the increase in phase separation (Table 3), the volume
of hard domain agglomerates increases (Fig. 2d and f), which is the result of reaction of NH$_2$ groups on nanosilica surface and NCO groups in hard segments. This reaction weakens chain mobility and creates favourable conditions for phase separation in this nanocomposite group.

4 Conclusion

Within the framework of the work the influence of the quantity and modification of nanosilica on the properties of polyurethanes was assessed; the polyurethanes are intended to be used as short-term implants exposed to abrasion wear. The addition of 0.5wt% of nanosilica caused a decrease in storage modulus and abrasion wear of nanocomposites. Bigger amounts of nanosilica result in a significant growth of storage modulus. The modification of nanosilica surface with NH$_2$ groups adds to the increase in the modulus and to a favourable fall in the abrasion wear of the examined nanocomposites.

The joined qualitative and quantitative analysis of AFM pictures of polyurethane structure and its nanocomposites as well as the analysis of phase separation degree, performed on the basis of FTIR spectra, permitted to explain the causes of changes in the properties brought about by the introduction of nanofillers.

Acknowledgements

The authors are grateful to the Warsaw University of Technology for financial support of this study. Thanks are also offered to Dr Maria Zielecka from the Institute of Industrial Chemistry for the carrying out of nanosilica modifications.

References


[22] Bliznyuk, V. N.; Tereshchenko, T. A.; Gumenna, M. A.; Tomza Y. P.; Shevchuk, A. V.; Klimenko, N. S.; Shevchenko, V. V. Polymer 2008, 49, 2298


