An experimental device designed to obtain repeatable condensation peaks in a closed system

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Abstract

The results of an experimental study related to repeatable condensation peaks in a closed system are presented and a discussion to describe their incipiencies is detailed. The test sections (D1: filling capacity = $15.14 \times 1e^{-9} \text{ m}^3$; D2: filling capacity = $6.45 \times 1e^{-9} \text{ m}^3$) were divided into three parts: an evaporator, an insulator and a conductive top cover where a pressure sensor was inserted. The impacts of several parameters have been studied: devices' filling ratio (α); hot temperature of the evaporator; cooling of the condensation chamber and; the devices' dimension. For all the obtained results the frequencies were low (< 0.15 Hz) but the peak pressure amplitudes were significant (up to 987 mbar). The influence of α had been studied and its increase led to an augmentation of the pressure amplitude (exponential-like function) but to a decrease in the oscillation frequencies. The hot temperature's increase enhanced the oscillation frequency but the pressure amplitude variations were slightly decreased. The cooling level diminution led to a frequency decrease, but to a slight variation of the pressure peak amplitude. Reducing the devices' dimensions from D1 to D2 for the same α and cooling level allowed the increasing of the oscillation frequency without significantly modifying the pressure peaks' amplitude. In a strategy for energy harvesting these oscillation peaks could be converted into electricity using piezo-electrics.

Keywords: oscillations, pressure peaks, energy harvesting, condensation *implosion*, evaporation, phase-change.



1 Introduction

The progress made in miniaturization has highly reduced the power required to work electronic systems and has opened the way to zero-power electronics (no batteries) [1]. One of the key technologies for the future smart systems concerns wireless sensors. In order to power these systems, energy harvesting from the environment has become a relevant challenge. Waste heat is one of the most significant sources of energy and trials for its conversion into electricity have begun. Until now, most of the well-known thermal harvesters are based on the Seebeck effect [2]. The two main drawbacks of Seebeck technology are: (1) the "good" thermal conductivity (generally due to good electrical properties) of the device which makes the maintenance of the thermal gradient difficult without the use of large heat sinks; (2) the need for rare materials.

Due to these drawbacks, new concepts have emerged to harvest waste heat [3–8]. Recently, many attempts for realizing MEMS comprising liquid/vapor phase change have been done. Most of the time the architecture remains simple and presents a planar shape with two diaphragms and a working liquid entrapped between them [3, 4]. One of the first prototypes presenting this architecture seems to be the P³ micro-engine realized by Whalen *et al.* [4]. The engine core was comprised between two diaphragms of silicon spaced at 60µm (Figure 1a). A vapor bubble of $3M^{TM}$ PF-5060DL was introduced between these diaphragms. Heating pulses were generated in order to drive (or not) the heat through the device. Oscillation frequencies of more than 400 Hz were reached in this study with an increase in typical pressures of 10 kPa.



Figure 1: a) Cross section of the prototype using the typical PCMHE architecture from [4]; b) oscillation due to phase change and bistable from [5]; c) micro-vibrations from [6].

Another technique (Figures 1b and 1c) using phase-change phenomena to also harvest energy has been reported. Huesgen *et al.* [5] used a bistable material in order to connect their device to the cold source, when vapor appears due to vaporization and connect it to the hot source when this vapor turns to liquid. For a temperature difference of 37° C the oscillation frequency reaches 0.71 Hz. All these studies usually used an additional oscillating media in order to conduct the heat for the oscillations' control. A promising system using a constant source of heat to produce energy had also been reported [1, 7, 8]. The system consisted of a bimetal, which can suddenly hit a piezo-electric element in order to produce energy. The sudden mechanism of snapping could be realized for different ranges of temperatures (depending on the materials used) and its origin comes from the shape of the bimetal. Puscasu *et al.* [8] put forward a scaling law which opened the way for the miniaturization of this technology.

The main goal of this study was to realize an oscillating system only controlled by a constant source of heat. These kind of oscillating conditions have already been obtained in literature. Usually the systems which used them are employed to transfer large amounts of heat. Indeed, a pulsating heat pipe (PHP) could achieve this oscillating behavior. These technologies have been studied since the 80s and already given promised results in electronics' cooling as shown in [9, 10]. The oscillation mechanisms, which occurred in pulsating heat pipes (that have a very simple architecture), are interesting. Indeed, a tube partially filled with a working liquid is used. The thermal boundary conditions to make the system work could be a constant source of heat and a cold source. The start-up of the system is initiated for a sufficient amount of heat regarding a specific filling ratio (α). The oscillations are due to heat and mass transfer between the liquid and vapor slugs. Several studies [9–13] have been conducted of the working of this technology and some general remarks can be given.

For tube diameters that are too low, Lin *et al.* [11] demonstrated that the heat flux needed for generate oscillations was increased (with water used as working fluid). Moreover for a tube diameter inferior to 0.4 mm the oscillations were not observed. These oscillations stop with diameter reduction were also observed by Lips *et al.* [12]. In order to make PHP work properly the authors calculated that the internal diameter of the tube needs to fulfill the following requirement:

$$d_h < C_{\sqrt{\frac{\sigma}{g(\rho_l - \rho_v)}}} \tag{1}$$

where d_h equals the hydraulic diameter, g, the constant of gravitation, σ the liquid surface tension, ρ the liquid or vapor density. The usual values of C \approx 2 are generally given.

Commonly the PHP loops have a serpentine shape, which comprised several turns. The numbers of turn could have an impact on the PHP performances. Lips *et al.* [12] showed that for less than 4 turns the thermal performances of a PHP could be decreased. Lin *et al.* [11] also demonstrated that when the length of the tube was smaller, the starting-up of the PHP was facilitated. Many other parameters also had an impact on the PHP working, such as: orientation (vertical or horizontal), the fact that PHP is open or closed, the working fluid, the filling ratio etc.



Khandekar *et al.* [13] demonstrated that with a simple set-up consisting of a single closed loop of capillary dimensions it was possible to obtain an oscillating behavior. The authors established that the oscillation of different frequencies in the range 0.1 to 3Hz could be achieved with such a device. Even though the frequencies are generally known in PHP the value of the pressure variation during these oscillations is usually not given.

Studies related to systems using liquid-vapor phase change, specifically in condensation situations, have also reported many sudden pressure variations [14, 15]. These phenomena are generally named: condensation induced water hammer (CIWH). Most of the time, this phenomenon appears for two-phase flow at a low flow rate and for stratified conditions. Due to the instabilities at the vapor/liquid interface a steam of vapor could be trapped between two liquid slugs and could collapse, leading to large pressure variations in the system.

Other condensation situations could lead to pressure variation. The mechanism of droplet grow rate has been studied by Rogers *et al.* [16] during heterogeneous vapor condensation. The authors found that the radius evolution of a condensing droplet was highly increased for coalescence conditions compared to an isolated condensing droplet (for a same time scale). In the present study specific oscillations due to a sudden decrease in pressure in a closed chamber were obtained in a configuration that has not yet been presented in the literature to the best of our knowledge. The oscillations were obtained for passive conditions (heat was not oscillating) and a parametric study is presented in order to better understand this mechanism.

2 Experimental setup and characterizing method

2.1 Set-up description and experimental procedure

The source temperature range was between 115° C and 145° C and the selected heated surface was 40 x 40 mm². The conditions for the environment were ambient. The tests sections are separated into three parts: the evaporator (made in stainless steel 316), an insulator (made in polyether ether ketone: PEEK) and a top cover (made in stainless steel 316) where a pressure sensor (KELLER 35 X HTC) was inserted. The mains thermo-physical of the materials used in this study have been reported in Table 1.

Material	ρ (kg/m3)	λ (W/m/K)	$C_p(J/kg/K)$
Stainless steel 316	8000	16.3	500
Peek	1320	0.245	2010

Table 1: Thermo-physical properties of materials.

Deionized water (not degassed) had been used as the working fluid. The main thermo-physical properties of the working fluid have been calculated (using REFPROP) and summarized in Table 2 for different saturation pressures.



WATER	$P_{sat} \approx 1 \text{ bar}$	$P_{sat} \approx 2 \text{ bar}$	$P_{sat} \approx 3 \text{ bar}$	$P_{sat} \approx 4 \text{ bar}$
	$T_{sat} = 100^{\circ}C$	$T_{sat} = 120.4^{\circ}C$	$T_{sat} = 133.7^{\circ}C$	$T_{sat} = 143.7^{\circ}C$
ρι (kg/m3)	958.37	942.77	931.69	922.78
$\rho_v (kg/m3)$	0.5976	1.1361	1.6577	2.1694
hlv (kJ/kg)	2256.4	2200.9	2163.0	2133.0
C _{pl} (kJ/kg/K)	4.2156	4.2442	4.2689	4.2913
C _{pv} (kJ/kg/K)	2.0799	2.1794	2.2641	2.3406
$\lambda_l (W/m/K)$	0.6790	0.6832	0.6835	0.6829
$\lambda_v (W/m/K)$	0.0250	0.0275	0.0292	0.0306
σ (mN/m)	58.917	54.884	52.174	50.072

Table 2: Thermo-physical properties of water for different saturation pressures.

In order to better understand the working of the oscillations, two geometries have been tested (Figure 2).



TEST SECTION 1 (D1) TEST SECTION 2 (D2)

Figure 2: Schematic of the test sections used with main dimensions.

The O-rings' grooves which allow to avoid gas or liquid leakage, are clearly visible in Figure 2. The main parameters for the main volumes and lengths are summarized in Table 3.

Test	h _{total}	hinsulator	Vtotal	Vcondensation	Vevap	Ratio
sections	(mm)	(mm)	(ml)	(ml)	(ml)	Vcond/Vevap
D1	31	15	15.14	12.38	2.64	4.68
D2	27	12	6.45	5.23	1.09	4.81

Table 3: Comparison of test sections geometry.

The main objective of this fabrication is based on the fact that the device must be built without any moving or rotary parts. Using this fabrication strategy the devices can be realized easily and with cheap materials. The experimental bench (Figure 3) comprised a heater from MINCO (HM6815: 12 Ω), a flux-meter from CAPTEC (linear, precision: 3%) which was also able to measure the temperature using a type-T thermocouple (hot temperature). The pressure variations were measured with a pressure sensor from KELLER (35 X HTC, full scale: 3bar; precision: 0.5%). Before each experiment, the flux-meter surface was cleaned using alcohol, dry with pressurized N_2 and then a thin layer of thermal grease (CW7270) was put on its two surfaces to ensure a good thermal contact with heater and test sections. Thereafter, the selected test sections were lain down on the flux-meter which was between the test section and the hot source. The heating power was regulated using a PID controller which was driven by a LABVIEW® program. The driving temperature was measured using a K-type thermocouple which was used to measure the cold temperature on the device top's surface (Figure 3). In order to cool down the test sections a fan was put above the top cover. The distance between the top cover and the fan was set at 12 cm.



Figure 3: a) Test section; b) experimental facility.

2.2 Study of specific parameters

As for PHP, some parameters were suspected to have an influence on the oscillations. For this study the parameters, which had been varied, are the:

- a. volume of water (filling ratio);
- b. value of hot temperature;
- c. cooling level.

3 Oscillations mechanism

During the experiments, for the specific range of filling ratio (α) the oscillations appear in the test sections when the heat flux was applied. The filling ratio can be expressed as following:

$$\alpha = \frac{v_l}{v_t} \tag{2}$$

where V_l is the liquid volume and V_t is the total volume available.

In order to check the leak-tightness, when the test sections were closed, the pressure variations were analyzed and the experiments were started only when its value remained stable. All our experimental points were analyzed for minimum time duration of 1 hour when the oscillating behavior was stabilized. The pressure and heat flux signatures usually evolved as shown in Figure 4.





Figure 4: a) Pressure; b) heat flux signatures for D1 ($V_l = 2000 \ \mu L$ and $T_{hot} = 130^{\circ}$ C). Mean pressure amplitude variation = 345.5 mbar; Mean heat flux variation = 900.0 W/m².

The pressure peaks amplitude (δP) and the heat flux variations ($\delta q''$) were synchronized (with a small lag) (Figure 4). All the oscillating signals were analyzed using the PWELCH method (with MATALB®) in order to determine the power spectral density and detect the peaks frequency.

According to our observations, the oscillating behavior could be detailed in following the steps, presented in Figure 5.



Figure 5: The working cycle of the oscillating device.

The oscillations were controlled by evaporation and condensation rates and specifically by the shape of the test sections. The pressure peaks were clearly due to condensation because of the orientation of pressure signature during the peak. These peaks were due to sudden condensation. As calculated by Rogers *et al.* [16] the radii of single droplets submitted to heterogeneous vapor condensation grows much slower than the radius of the equivalent droplet resulting from the coalescence of multiple droplets in condensation.

By considering the vapor volume, the oscillating cycle comprised four steps. In our test section the behavior could be expressed as the following: 1) after the pressure peak, a compression of the vapor occurred during which the temperature decrease; 2) the pressure reaches a "plateau" (a small time during which the condensation rate and the evaporation rate have almost the same value); 3) after this step the pressure decreases slowly because of a lower evaporation rate. Indeed due to the geometry all the condensed liquid does not return immediately to the evaporator. During the condensation process small droplets are in formation on the cold surface (pressure sensor surface) and their diameters increase slowly; 4) the droplets on the cold surface could suddenly coalesce and the pressure decrease sharply due to the fact that more vapor turns into liquid. Then, due to this sudden pressure decrease, the rate of vaporization of the liquid, in contact with a hot surface, highly increase leading to a pressure increase. During this time a significant increase in the heat flux (an also a decrease in hot temperature) have been observed.

4 Results and discussion

4.1 Influence of volume on oscillations

For D1 and D2 a clear variation of the oscillation had been observed with α .

When α was increased, the amplitude of the pressure drop variations increased. On the contrary the frequency generally decreased with α . This behavior could be explained by the fact that the condensation process is slowed down when the liquid volume is increased. Indeed, the mean pressure in the system increases with α . Owing to this pressure evolution, the temperature at which the condensation occurred also increased leading to a decrease in the condensation rate for the same cooling conditions. Indeed, the heat transport toward the cold surface was slowed down and the cold temperature generally decreased with α for same boundary conditions (Figure 6).



Figure 6: a) Volume impact on pressure amplitude and oscillations for D1 and D2 at $T_{hot} = 130^{\circ}C$; b) Impact of filling ratio on heat flux and ΔT .

Below 1000µL (D1, $\alpha = 6.61$ %) or below 800µL (D2, $\alpha = 12.41$ %), the oscillations were not fully continuous and a specific behavior with stopovers had been observed (Figure 7). This effect could be explained by the fact that water could, for low α , totally evaporate on the hot surface. Even though liquid is still present in the system (*partial pressure is not sufficient to suspect that water totally turns to vapor*) it can be assumed that during a period the liquid droplets were stuck in the condenser part. Indeed in order to return back to the evaporator the pressure head of the liquid should overcome the capillarity pressure:

$$\rho_l g H > \frac{2\sigma \cos(\theta)}{r} \tag{3}$$



where g is the gravitational constant, θ the contact angle between liquid and PEEK (around 110° for water at 25°C) and r the capillary radius. Indeed in order to fulfill the conditions of eqn 3, H > 1.216 mm.

During these waiting times, no oscillations occurred in the device. Nevertheless due to the fact that oscillations continued after theses bulges, a peak frequency was obtained at 0.1163 Hz here. All the volumes for which these waiting times were observed are circled and the respected percentage of time oscillation duration had been written near the corresponding values.



Figure 7: Visualization of the oscillations (and stopovers) for D2 and for $\alpha = 12.41\%$ for T_{hot} = 130.6°C; Peak frequency = 0.1163 Hz; Mean pressure variation = 177.7 mbar.

The influence of the dimensions on oscillations was analyzed (Figure 6). Indeed, it was possible to observe that with the reduction of the test section (D1 to D2) the peak oscillation frequency shift toward a higher frequency, whereas the pressure amplitude decreased as presented in Figure 6. This was due to the fact that for a smaller equivalent hydraulic diameter (D2 compared to D1) but the same heat flux, the vaporization rate is higher.

4.2 Influence of hot temperature

The hot temperature impact was also analyzed in this study. The range of the studied temperature was set between 115°C and 140°C (Figure 8a).



Figure 8: a) Pressure signature evolution with increase of temperature for D2;
b) impact of hot temperature on oscillations behavior α = 12.41% for D2 (V = 800µL).

As temperature increases the oscillating frequency also highly increase (from 0.012 Hz at 115°C to 0.147 Hz at 137°C). In the same time the bulges (see Figure 7) get more amplitude. Nonetheless even with these stops, oscillation duration time last more than 73% of the time when $T_{hot} > 125$ °C. Thus, the hot temperature increase had positive impact (increase in vaporization rate) on oscillation frequency. In the other hand, pressure variation amplitude decreased slightly.

4.3 Influence of cooling

In order to estimate the influence of cooling on the behavior of the system, the fan velocity was decreased gradually. The heat exchange coefficient was calculated using:

$$h_{conv} = \frac{q_{loss}^{\prime\prime}}{s(T_{cold} - T_{amb})} \tag{4}$$

The evolution of pressure, hot and cold temperatures and heat flux signatures are given in Figure 9.



Figure 9: a) Pressure; b) temperatures and heat flux signatures for cooling reduction for D2 ($\alpha = 12.41\%$ and $T_{hot} \approx 120^{\circ}$ C).

The influence of cooling diminution was studied on the D2 test section for two different hot temperatures. The results are presented in Figure 10.

It is possible to observe that for $T_{hot} \approx 130^{\circ}$ C, the oscillations were not fully continuous (Figure 7) whereas for $T_{hot} \approx 120^{\circ}$ C the oscillations were always continuous for the all cooling range. The analyses of the behavior for $T_{hot} \approx 130^{\circ}$ C showed that the oscillating behavior was stabilized for a heat exchange coefficient between 25 and 35 W/m²/K.

When a sufficient heat exchange was attained the oscillations stabilized and further cooling does not improve the oscillations behavior very much (frequencies for $T_{hot} \approx 130^{\circ}$ C). In the same manner the pressure amplitude variations changed very slightly with cooling reduction.



Figure 10: Influence of cooling on D2 for two different Thot ($\alpha = 12.41\%$).

5 Conclusions

Oscillating test sections using sudden condensation phenomena have been fabricated in this work. These devices were separated into three parts: an evaporator which is in contact with a constant hot source, a thermal insulator which allowed the maintaining of a convenient temperature difference between hot and cold sources and a top cover in order to mount a pressure sensor for the pressure measurements. Pressure, heat flux variations and the temperature of cold and hot surfaces were analyzed in this study. The reasons for the appearance of the oscillations have been also given. Indeed, the selected devices' construction was one the main reason leading to their incipience. A parametric study had also been realized and the parameters which have been varied are: liquid filling ratio, hot source temperature, cooling level and device dimensions (two test sections were fabricated). The observed frequencies were in the range of 0.01 Hz to 0.15 Hz, and the pressure amplitudes were in the range of 94.8 mbar to 987.1 mbar. The results had shown that the device reduction from D1 (filling capacity = 15.14x $1e^{-9}$ m³) to D2 (filling capacity = 6.45 x $1e^{-9}$ m³) improved the oscillation frequency and slightly diminished the pressure amplitude variations for the same filling ratio, heating and cooling conditions. The increase of these oscillation frequencies could open the way to energy harvesting by using this pressure drop to generate repeatable strains on a piezo-element.

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