

Influence of geochemical and anthropogenic sources on rainwater chemical composition in two coastal sites impacted by the gas and oil industry in Campeche, Mexico

R. M. Cerón¹, J. G. Cerón¹ & M. Muriel²

¹*Chemistry Faculty, Autonomous University of Carmen, Mexico*

²*Mexican Petroleum Institute, Marine Zone Direction, Mexico*

Abstract

The purpose of this research was to study rainwater chemical composition to assess the impact of marine aerosol and anthropogenic emissions, using Cl⁻ as a sea-salt aerosol tracer. Influence of different sources to excess of ions was detected by means of back air-mass trajectories. Fieldwork was done from July to November 2004, in two coastal sites, Carmen Island and San Antonio Cardenas in Campeche Gulf, Mexico. Rainwater samples were analyzed for Na⁺, Mg²⁺, K⁺, Ca²⁺, Cl⁻, SO₄²⁻, NO₃⁻ and NH₄⁺, the average concentrations are reported and the results discussed. The observed concentrations are compared with other works in the Caribbean Sea Region and are interpreted in terms of enrichment factors and meteorological conditions.

Keywords: rainwater chemistry, marine aerosol, coastal rainwater composition, Mexico.

1 Introduction

Background level is a useful tool when a diagnosis about the possible environmental impact caused in a specific zone is required, allowing to establish a comparison when reference laws or standards of non criteria pollutants are not available. Commonly, these values are established for remote areas with a minimal anthropogenic influence, by this way, it is possible to accomplish a diagnosis of impact caused by non criteria pollutants such as rainwater acidity, carbonyl compounds, and so on. In fact, research on precipitation chemical



composition of probably impacted zones, allow us to identify, first if there is an acidity problem, and second, to identify the main sources contributing to this acidity, and then to understand the process of transformation, transport, dilution and deposition of air pollutants. Objectives are accomplished by means of correlation between elements, local meteorology and back air-mass trajectories. Therefore to assess the impact from geochemical sources (marine aerosol, volcanic activity, and crustal) on chemical composition of rainwater in a specific region, knowledge of the background levels of wet deposition is required.

An important feature of the environmental assessment studies is the identification of all possible sources (biogenic, geochemical, natural, and anthropogenic) and to determine their relative contribution to the mineralization of rainwater.

Nriagu and Pacina [15] reported that human activities have an important effect on global and regional cycles of most of trace elements, whereas Duce *et al* [8] established that geochemical sources also may have an important influence upon trace elements through the different steps integrating biogeochemical cycles of elements in a region.

For this reason, it is essential to discuss the origin of trace elements to identify main sources using tracers. A method commonly used by geochemists to relate an element in rainwater with its geochemical source, involves the use of a tracer derived from the specific reference source. Therefore, to assess the enrichment of an element relative to its source, it is necessary to define the excess (fraction with an origin different from reference source), in terms of enrichment factor (EF). The most important geochemical sources contributing to rainwater mineralization in coastal and marine sites, where volcanic activity is not present, are marine aerosol and crustal, where sodium and chloride; and aluminum, respectively, are the main tracers, and where the specific reference sources are sea water and particles of crustal or soil, respectively (Chester *et al.* [5]).

Chemical composition of sea water is expressed as the ratio between the concentration of the element and the concentration of chloride ion, and it has been reported elsewhere (Culkin and Cox, [7]; Morris and Riley, [13]). In coastal and marine sites, it is assumed that relative proportion of elements as Na^+ , K^+ , Ca^{2+} , Mg^{2+} , Cl^- and Br^- in rainwater presents similarity to that reported for sea water. However, in some coastal and marine sites, the ratios of these elements are different in a significant way from their ratio for sea water, in spite of, it is believed that sea-salt is the main source of these trace components.

To assess the significance of sea surface as a source of aerosols, commonly, enrichment factor (EF) is used as an effective tool. An EF approaching to unity suggests that the trace element is originated in sea-salt (no enriched elements), whereas an EF greater than unity, suggests that a significant portion of the trace element has origin in a source different than sea-salt (enriched elements).

However, in the most of urban and industrial sites, trace elements atmospheric concentrations such as sulfate, zinc, cadmium and lead are significantly greater than those expected from the dispersion process of marine and mineral (crust) aerosols. Human activities are the main cause of the

enrichment of these elements, and often they are referred as enriched elements or excess, in this case, local or regional meteorology data are necessary to identify the main responsible sources. Therefore, the excess of a constituent in a rainwater sample represents an increase of the element upon the levels expected from sea water and crustal particles and allow us to infer by using meteorology data, the influence of another sources such as anthropogenic sources on chemical composition of rainwater.

There is a clear increase in the contribution of nitric and sulfuric acids to rain acidity in the Gulf of Mexico, as air mass move toward the west (Bravo *et al*, [2]). Parungo *et al* [16] reported enrichment factors for SO_4^{2-} of 10, and high levels of NO_3^- in the Gulf of Campeche. However, Muriel *et al* [14] concluded in according to meteorological conditions prevailing in the zone, that air pollutants released from offshore platforms for exploration and production of gas and oil, are transported and deposited on Tabasco coastline. On the other hand, Ahmed *et al* [1] report that petroleum facilities in the east coast of Arabian affect sulfate and nitrate levels in rainwater, where both, nitrate and sulfate contribute in equal proportion to the acidity.

The present paper reports ions measurements data of wet deposition for rainy season in 2004, in two locations: one in a semi-urban marine site (Carmen Island), and the other, in a coastal site presumably impacted by the gas and oil industry (San Antonio Cardenas), both located in Campeche State, Mexico.

2 Experimental

2.1 Sampling sites description

Development of petroleum industry has been accelerated in the last years, in the region of Atasta-San Antonio Cardenas located in Campeche State. In this region, there is a gas recompression plant, where sour gas from offshore platforms is sent to petrochemical facilities to be processed. This recompression plant has four flare burners that use sour gas as fuel, releasing pollutants to the atmosphere. These flare burners or vent system compose the safety system of the plant avoiding over-pressure in transport pipes.

Both sampling sites (Carmen Island and San Antonio Cardenas) are into the Natural Reserved Area named "Laguna de Terminos" (Figure 1). Carmen Island is surrounded by Laguna de Terminos, at $18^\circ 40'$ N latitude and $91^\circ 52'$ W longitude, 40 km away from San Antonio Cardenas and 30 km away from recompression plant of sour gas in Atasta.

San Antonio Cardenas is located in Atasta Peninsula at $18^\circ 37'$ N latitude and $92^\circ 13'$ W longitude, 40 km away from Carmen Island and 10 km away from recompression plant of sour gas in Atasta.

Pom-Atasta Lagoon system is 30 km away from Carmen Island, at $18^\circ 33'$, $18^\circ 38'$ N latitude and $92^\circ 01'$, $92^\circ 14'$ W longitude.

Laguna de Terminos has a total surface of 705, 016 Ha and is located at $18^\circ 03' 27''$, $19^\circ 10' 35''$ N latitude and $92^\circ 28' 38''$, $91^\circ 03' 16''$ W longitude.



Mean annual temperature in this region is of 27.2 °C, with rains occurring along the summer. Mean annual pluvial precipitation is of 1500 mm/year, and prevailing winds blow from NE and SE almost all the year with average velocities of 1.4 m/sec.

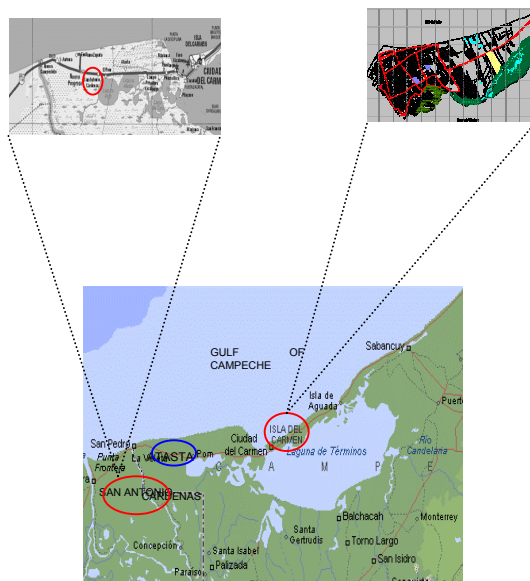


Figure 1: Map of the sampling sites location: San Antonio Cardenas and Carmen Island, identifying the location of the main emission source.

Campeche community, Natural Reserved Area committee and citizens of Atasta and San Antonio Cardenas have shown, since several years ago a constant concern about corrosion and its effects on their houses' roofs (mainly made of metallic sheets) and also about the effects caused on mangrove, and Pom-Atasta-Terminos lagoon system; there is also a possible effect on fisheries because it is their main way of living.

2.2 Sampling and analysis methodology

Precipitation samples in both sites were collected using two automatic wet deposition collectors (Tisch, Inc) from 1 June to 30 October 2004. In San Antonio Cardenas, one collector was installed on the roof of a building in Technological University of Campeche. The other collector was installed on the roof of a building in Autonomous University of Carmen in Carmen Island. Trained technicians checked the samplers daily, typically between 8:00 and 10:00 h, local standard time, to retrieve, preserve with chloroform and refrigerate any collected precipitation. Samples with volume lesser than 250 ml or polluted were discarded.



Samples were sent weekly to Environmental Sciences Laboratory at UNACAR (Autonomous University of Carmen), where they were filtered (with a Kontes Ultra-Ware System, using a vacuum pump and Phenomenex membranes) and analyzed for NH_4^+ , NO_3^- , SO_4^{2-} , Cl^- , Na^+ , K^+ , Ca^{2+} and Mg^{2+} .

Chloride, nitrate and sulfate were analyzed by non-suppressed chromatography with an Alltech chromatograph. For ammonium ion, an UV-Visible spectrophotometer (Perkin Elmer) was used for its determination by a colorimetric method. Na^+ , K^+ , Ca^{2+} and Mg^{2+} were analyzed by flame atomic absorption spectrometry with a GBC atomic absorption spectrophotometer. pH and conductivity were also measured by using a potentiometer (HACH model EC10) and an YSI 3200 conductivity instrument. Ion balance and conductance percent differences of data were used for quality assurance. Also, field blanks were analyzed to guarantee the cleanness of the sampling material.

2.3 Meteorology

Yucatan Peninsula shows two climatic periods well defined, a dry season (From November to May) and a rainy season (From June to October). Sampling period occurred during summer and autumn; therefore it was not possible to assess seasonal patterns for each ionic component present in the rainwater of the two sampling sites. In both locations, surface winds data were not available. In this case, to identify the possible sources different from marine or mineral (crust) aerosol contributing to the excess levels or enrichment of ionic components in rainwater, back-air mass trajectories were used. The specific location of the site, the date and time of each precipitation event were registered; this data were provided to the NOAA Hybrid Single Particle Lagrangian Integrated Trajectory (HYSPPLIT) model to generate 24-hour back trajectories. Examples of typical HYSPLIT trajectories are illustrated in Figure 2.

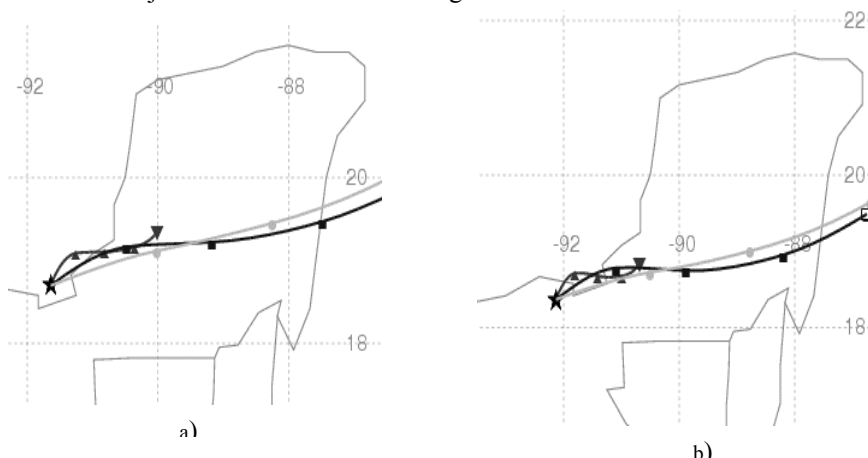


Figure 2: Illustration of 24-hour air mass back trajectories used to identify the main pollution sources for two sampling sites: a) Carmen Island and b) San Antonio Cardenas.

From figure 2, it can be observed that in both sites, prevailing winds during sampling period had an east component, mainly NE. San Antonio Cardenas was influenced by local anthropogenic sources located at NE from sampling site (Sour gas recompression plant), whereas Carmen Island was under the influence of maritime air and any significant anthropogenic influence was not identified.

3 Results and discussion

Ionic abundance obtained in San Antonio Cardenas and Carmen Island is presented in Table 1.

Table 1: Ionic abundance obtained in San Antonio Cardenas and Carmen Island.

Elements	San Antonio Cárdenas	Carmen Island
Cations	$\text{Na}^+ > \text{Ca}^{2+} > \text{K}^+ > \text{Mg}^{2+} > \text{NH}_4^+ > \text{H}^+$	$\text{Na}^+ > \text{Ca}^{2+} > \text{Mg}^{2+} > \text{K}^+ > \text{NH}_4^+ > \text{H}^+$
Anions	$\text{SO}_4^{2-} > \text{Cl}^- > \text{NO}_3^-$	$\text{Cl}^- > \text{SO}_4^{2-} > \text{NO}_3^-$

The most abundant cations in both sites were Na^+ and Ca^{2+} , probably due to influence of marine aerosol and crust (because of calcareous soils are abundant in Yucatan Peninsula). Agricultural practices are not present in any of two sites, for this reason NH_4^+ levels were very low. The most abundant anions in two locations were Cl^- due to influence of sea-salt aerosol, and SO_4^{2-} , these high sulfate levels suggests there is influence of sources different from marine aerosol, mainly in San Antonio Cardenas. Table 2 shows the volume-weighted mean concentrations (VWM) for all collected samples in both sites.

Table 2: Volume-weighted mean concentrations in mg l^{-1} for wet precipitation collected in San Antonio Cardenas and Carmen Island.

Sampling site	Na^+	K^+	Ca^{2+}	Mg^{2+}	SO_4^{2-}	Cl^-	NO_3^-	NH_4^+
San Antonio Cardenas N=26	2.47	0.37	1.06	0.33	3.85	3.84	2.50	0.09
Carmen Island N=26	3.06	0.31	0.98	0.37	1.98	5.06	0.90	0.11

The rainwater chemical composition observed in both sampling sites (Na^+ , K^+ , Ca^{2+} , Mg^{2+} , and Cl^- , except for nitrate and sulfate) was similar to that observed in other coastal and marine sites in the Caribbean Sea by other researchers (Cerón *et al* [4]; Bravo *et al* [2]; Clark *et al* [6]; Eklund *et al* [9]; Hendry *et al* [11]; Mc Dowell *et al* [12]).

Based on enrichment factors (EF) results, it can be concluded that Na^+ , Mg^{2+} and Cl^- have an origin typically marine (EF close to 1), whereas K^+ , Ca^{2+} and SO_4^{2-} are enriched elements and showed a significant excess (EF>1), signifying that other sources are contributing notably to their levels present in rainwater. Sea-salt aerosol contributed to 91 % of Na^+ and 90% of Mg^{2+} in Carmen Island; and 86% of Na^+ and 79% of Mg^{2+} in San Antonio Cardenas, suggesting that the influence of marine aerosol is more limited in San Antonio Cardenas. Since chloride ion was taken as the basis for excess calculations, sea-salt contributed to 100% of Cl^- in both locations.

Proportions of sea-salt for K^+ , Ca^{2+} and SO_4^{2-} were lower in both sampling sites, contributing to 34% of K^+ , 11% of Ca^{2+} and 36% of SO_4^{2-} for Carmen Island, and 21% of K^+ , 7% of Ca^{2+} and 14% of SO_4^{2-} for San Antonio Cardenas, respectively.

High sulfate levels in both sites are comparable to that found in sites with anthropogenic influence. The average excess sulfate concentration (SO_4^{2-})_{xs} based on chloride was 69.1 and 26.43 $\mu\text{Eq l}^{-1}$ for San Antonio Cardenas and Carmen Island, respectively. These values are almost seven and three times higher than (SO_4^{2-})_{xs} background hemispheric value reported by Galloway *et al* [10] for marine remote sites (10 $\mu\text{Eq l}^{-1}$), and it can be observed that anthropogenic influence in San Antonio Cardenas is greater than that observed in Carmen Island.

High nitrate concentrations were found in both locations, 11.06 and 3.98 $\mu\text{Eq l}^{-1}$ for San Antonio Cardenas and Carmen Island, respectively. These values exceeded the NO_3^- background hemispheric concentration proposed by Casimiro *et al* [3] for coastal and marine remote sites (2.8 $\mu\text{Eq l}^{-1}$), mainly in San Antonio Cardenas, suggesting the direct influence of anthropogenic sources in this site.

On the other hand, low ammonium levels were found in both sampling sites, this means that local sources with a significant contribution were not available, since agricultural practices are not present.

Even in areas considered as free of pollution, pH value for natural rainwater is of 5.6 due to presence of CO_2 that dissolved in water to produce carbonic acid, the main responsible of this value. When pH value ranged from 5 to 5.6, it can be considered that rainwater has enough buffer capacity so it is not an impacted site; but, when pH value is less than 5, there is an evident anthropogenic source contributing to the acidity of rainwater. pH values in this study varied from 3.01 to 6.27 in San Antonio Cardenas and from 2.98 to 6.91 in Carmen Island. Mean pH value for San Antonio Cardenas was 4.64 and for Carmen Island was 5.38, suggesting a direct influence anthropogenic over chemical composition of rainwater in San Antonio Cardenas, therefore, this site can be considered as an impacted site.

The contribution of strong acids to the acidity of rainwater was determined from equivalent (SO_4^{2-})_{xs}/ H^+ and NO_3^-/H^+ ratios; in both sites, these ratios were greater than unity, signifying that both acids contributed to pH values observed, being (SO_4^{2-})_{xs} who contributed in a greater proportion than nitrate, mainly in San Antonio Cardenas.

Correlation coefficients were calculated using a Pearson's matrix, significant correlations ($r^2 > |0.75|$) for main ions present in rainwater at San Antonio Cardenas and Carmen Island are shown in Table 3.

Table 3: Correlation coefficients among main ions present in rainwater at San Antonio Cardenas and Carmen Island.

Ion pairs	Carmen Island	San Antonio Cardenas
$(K^+) - (Ca^{2+})$	0.9423*	0.9805*
$(Na^+) - (Cl^-)$	0.9386*	0.9985*
$(Na^+) - (Mg^{2+})$	0.9489*	0.8914*
$(pH) - (SO_4^{2-})$	-0.9535*	-0.8450*
$(SO_4^{2-}) - (NO_3^-)$	0.9625*	0.8356*
$(pH) - (NH_4^+)$	0.8863*	0.9081*

* Significant at the 0.005 level.

In both sites, elements showed a common trend; K^+ and Ca^{2+} had significant correlation, indicating a common source, probably particles of calcareous soils (crustal). On the other hand, a high correlation between $(Na^+)-(Cl^-)$ and $(Na^+)-(Mg^{2+})$ indicates a marine origin since these elements are typically marine. In both cases, the concentrations of NO_3^- and SO_4^{2-} demonstrate significant correlations, indicating their relationship to common anthropogenic sources and transport processes. pH had a significant inverse correlation with SO_4^{2-} for both sites, this represents that SO_4^{2-} contributes in a great proportion to the acidity of rainwater; whereas NH_4^+ is positively well correlated with pH, indicating that it can play a significant role in neutralization process of strong acids.

4 Conclusions.

In this paper the results of one year rainwater chemistry study were presented for a coastal site and an island, both located at Campeche Coast in Mexico. Excepting NO_3^- and SO_4^{2-} , ionic concentrations are in agreement with reported values in the Caribbean Region (Cerón *et al* [4]; Bravo *et al* [2]; Clark *et al* [6]; Eklund *et al* [9]; Hendry *et al* [11]; McDowell *et al* [12]). Marine aerosol contributes in a great proportion to Na^+ , Mg^{2+} and Cl^- in both sites, whereas K^+ and Ca^{2+} showed enrichment and probably are derived from crust. On the other hand, NO_3^- and SO_4^{2-} levels in both locations, exceeded background levels reported for coastal and marine remote sites, suggesting that sources different from sea-salt aerosol are contributing to the mineralization process of rainwater, being this contribution more significant in San Antonio Cardenas. An analysis of air mass back trajectories indicates that main source for these high levels of NO_3^- and SO_4^{2-} is the sour gas recompression plant located at NE from San Antonio Cardenas. Results of this study made possible an environmental diagnosis and it can be concluded that Carmen Island is not impacted, whereas San Antonio Cardenas showed a strong acidity in the rainwater and it is directly impacted



from anthropogenic activities, specifically the sour gas recompression plant located 10 km at NE.

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