

# Impact of motorcycles on urban tropospheric ozone

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## Abstract

The increased number of motorcycles in the large Brazilian cities is due to several factors such as traffic, low cost, mobility, few parking lots and low efficiency of public transportation, making it an important factor in air quality deterioration. In this context, vehicle emissions monitoring is essential to understanding the contribution on air pollution. Emissions of two representative motorcycles using gasohol (gasoline with 22% of ethanol) were evaluated on a dynamometer test bench and analyzed by gas chromatography mass spectrometry and on-line analyzers according to the EC/97/24 standard. Emission data obtained in combination with meteorological data and ambient air pollutants for the city of Rio de Janeiro (Brazil) during the winter of 2011, using the trajectory model OZIPR and chemical model SAPRC to assess the impact on the ozone forming potential in the troposphere of Rio de Janeiro city for the next years. The results indicate that ozone levels will exceed the established limits by national legislation within three years. The increase in ozone concentration occurs due to high emissions of reactive volatile organic compounds in an atmosphere with high levels of nitrogen oxides. Given this scenario, additional measures are necessary for the management of emissions from mobile sources.

*Keywords:* atmosphere, modelling, motorcycle, ozone, troposphere, VOC.

## 1 Introduction

Products of internal combustion engine vehicles can be considered one of the most polluting activities, mainly due to the large number of vehicles in cities.



The low quality and quantity of public transportation is one of several reasons for vehicle fleet growth in metropolitan regions. According to the latest report on air quality for the state of São Paulo (Brazil), vehicles are responsible for 97% of CO, 77% of HC emissions, 82% of NO<sub>x</sub>, 36% of SO<sub>2</sub> and 40% of PM. In addition, motorcycles contribute 16% of the total CO and 13% of the total HC [1]. The high emission factors for such vehicles is due to the absence of electronic fuel injection, catalysts and anti-evaporation systems

Motorcycles are a rising form of transportation in large cities across the world, especially in emerging countries such as China, India and Brazil. The intensive use of motorcycles is explained by their high mobility in increasingly congested cities that lack affordable public transport, as a motorcycle with a less powerful engine (150 cc) usually costs 20 to 25% that of a small car and is also associated with low fuel consumption, ease of parking and low maintenance cost. These advantages outweigh the inherent dangers of this form of transportation and other disadvantages such as interfering rain and the high cost of insurance. Brazil recorded 428,970 traffic accidents in 2008; the number of vehicles involved was 597,786, of which 246,712 were cars and 200,449 were motorcycles [2].

Given the facts presented and the potential increase in the amount of hydrocarbons in the atmosphere due to motorcycles, the net effect of ozone formation in cities such as Rio de Janeiro is also enhanced by high NO<sub>x</sub> levels. To perform simulations of tropospheric ozone formation from motorcycle emissions, the trajectory model OZIPR (Ozone Isopleth Package for Research) coupled with the chemical model SAPRC (State-wide Air Pollution Research Centre) was used [3].

This study aimed to assess the overall impact of motorcycle emissions on air quality in large cities such as Rio de Janeiro by studying the profile of current fleets and estimates for future scenarios and identifying their issues in classes of compounds.

## 2 Experimental

### 2.1 Motorcycle emissions sampling

Samples were collected during emissions testing in a dynamometer bench (AVL Zöllner model AN 40 770, 648 mm, 100 kW). Two motorcycles were tested (150 cm<sup>3</sup>, and 125 cm<sup>3</sup> engine displacement, respectively named BK01 and BK02, respectively). Both were Euro III models with EFI, catalyst, and 2009 model. The exhaust emissions were measured using commercial gasoline with 22% v/v of ethanol.

Tests were conducted over a chassis dynamometer according to the 97/24/EC drive cycle, which is divided into six primary modules and one extra-urban cycle. Regulated emissions (CO, HC and NO<sub>x</sub>) have been measured with 7000 series Horiba benches. Each cycle includes four modes (idling speed, acceleration, constant speed, deceleration etc.). Throughout the test, a constant flow (diluted 1:20) to a 90 L Tedlar bag was employed to successively determine

the concentrations of carbon monoxide (CO), hydrocarbons (HC) and nitrogen oxides (NO<sub>x</sub>), which were then correlated to the distance travelled. A test procedure layout can be observed in Figure 1.

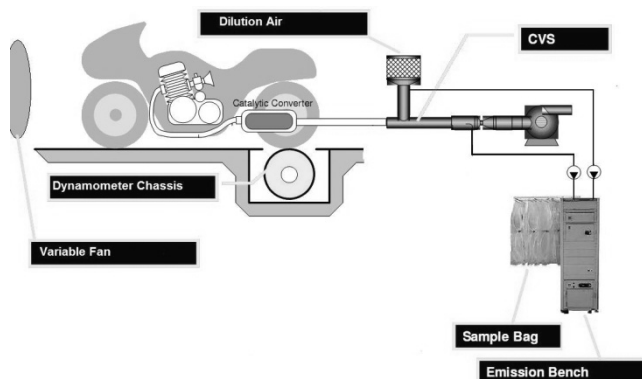


Figure 1: Test procedure layout.

For Volatile Organic Compound (VOC) sampling and speciation, 6 L electropolished stainless steel canisters (SUMMA) were used.

## 2.2 Analysis of criteria pollutants and VOC speciation

The analytical methodology used for the VOC speciation followed the TO-15 USEPA methodology [4] with two gas chromatographs: one to analyse the hydrocarbons from C2 to C5 and one to analyse them from C6 to C12.

The VOC analyses were performed by passing a known volume through a cryogenic pre-concentrator trap (-180°C) and then injection in a DB-1 60m column, 0.32mm i.d. with 1.0 µm film. The temperature was set between -50°C and 180°C, with heating at a rate of 6°C min<sup>-1</sup>. The sample was divided into two different detectors: mass spectrometer, where the identification of compounds was made, and a flame ionization detector (FID), where the quantification was performed, using a calibration curve with the NPL 30 EU Directive Ozone Precursor Mixture component at 20 ppb.

Carbonyls samples were collected and analysed following the US EPA TO-11A methodology [5], based on the reaction of atmospheric carbonyls with 2,4-dinitrophenylhydrazine, to form the respective hydrazones. The samples were collected using two silica-C18 cartridges in series impregnated with an acid solution of 2,4-dinitrophenylhydrazine (DNPH), using an air pump (KNF UNMP 850 KNDC) operated at 1.0 mL min<sup>-1</sup> for 60 minutes. An ozone trap filled with potassium iodide was used before the DNPH cartridge to avoid ozone interference. After the sampling, the cartridges were sealed, wrapped with aluminium foil, stored at 4°C and analysed after 3 days. The detailed procedure is thoroughly explained in our previous work [6–8].

The extraction of carbonyl samples was performed using 5 mL of acetonitrile and chemical analyses were completed with high-performance liquid chromatography (HPLC), using a Perkin Elmer Series 200 equipment with UV detection at 365 nm. The mobile phase used was 55% acetonitrile in water at 30°C, with a 30  $\mu\text{L}$  volume and two C18 columns in series (250 mm, 4.6 mm and 5.0  $\mu\text{m}$ ). Calibration was conducted using a standard mixture (Supelco CARB Carbonyl Mix 1) containing formaldehyde, acetaldehyde, acrolein, acetone, propionaldehyde, butyraldehyde and benzaldehyde at concentrations between 0.5 to 10.0  $\text{mg L}^{-1}$ , yielding correlation coefficients higher than 0.99.

### 2.3 Model development

The modelling procedure was performed using the trajectory model OZIPR (Ozone Isopleth Plotting Package), which is widely accepted and well documented in the literature [9, 10]. Other publications by our group detail its applicability [11, 12]. In summary, the input data for a simulation are the following:

- Speciation of VOC of the local atmosphere in ppmC fractions.
- Hourly emissions of total VOC, NO<sub>x</sub> and CO in  $\text{kg km}^{-2} \text{h}^{-1}$ ;
- The initial concentrations of total VOC, NO<sub>x</sub> and CO in ppm;
- Coefficients of wet and dry deposition for the main pollutants.
- Hourly weather data for temperature, pressure, humidity and mixing height;
- The actinic solar flux, used to calculate the intensity of sunlight;
- A chemical model with chemical reactions and their rate constants.

The output data are the hourly average concentrations of the main chemical species (e.g., NO<sub>x</sub>, CO, CH<sub>4</sub>, O<sub>3</sub>, and PAN) and secondary pollutants isopleths for several values of VOC and NO<sub>x</sub>.

The chemical model used was the 1999 version of the SAPRC model (Statewide Air Pollution Research Centre) developed by Carter; several updates were added in later years [3, 13–18].

Ambient air criteria pollutants data (CO, NO<sub>x</sub>, NMHC, O<sub>3</sub>) were collected by an automated monitoring station during the winter of 2011, located at the south part of the city of Rio de Janeiro, Brazil (22°58'44"S 43°13'58"W), which also measured meteorological parameters (temperature, pressure and humidity). The location is adjacent to a large parking lot and is one of the busiest traffic routes in the city. It is also located next to the largest urban forest in the world, the Tijuca Forest.

The samples for VOC were collected near the automated monitoring station during the winter (August and September 2011). Samples were collected in the first morning hours when the sunlight is at a minimum and the presence of secondary pollutants can be neglected.

Canister samples were collected every five minutes and two samples were collected for three days. Samples were collected from 6:45 AM to 7:15 AM on August 29, 6:40 AM to 7:05 AM on August 30 and 7:15 AM to 7:30 AM on September 6.

### 3 Results and discussion

According to the INEA report [19] 321,000 tons of CO, 90,500 tons of NO<sub>x</sub> and 79,300 tons of HC are emitted each year in the MRRJ, which has an urban area of 2,400 km<sup>2</sup>. To obtain the hourly emissions calculations, a year was considered to have 313 days, discounting Sundays and holidays, due to low circulation of vehicles on these days. For each day, 18 hours of effective movement of vehicles were considered. Hourly emissions estimates were 6.69 kg NO<sub>x</sub> h<sup>-1</sup> km<sup>-2</sup>, 23.7 kg CO h<sup>-1</sup> km<sup>-2</sup> and 5.87 kg VOC h<sup>-1</sup> km<sup>-2</sup>. These values were loaded into the model to reproduce experimental hourly concentrations of CO, NO<sub>x</sub> and O<sub>3</sub> provided by automatic monitors for August 29, 2011. The results for the VOC speciation are detailed in Table 1.

Table 1: VOC speciation for August 29, 2011.

| VOC                       | Sample 1 | $\mu\text{g m}^{-3}$<br>Sample 2 | Average |
|---------------------------|----------|----------------------------------|---------|
| Formaldehyde *            | 10.34    | 13.2                             | 11.77   |
| Isobutane                 | 0.04     | 0.08                             | 0.06    |
| Acetaldehyde *            | 3.18     | 4.66                             | 3.92    |
| Acetone *                 | 3.80     | 4.02                             | 3.91    |
| Propanal *                | 1.12     | 1.27                             | 1.19    |
| 1-butene                  | 2.01     | 3.09                             | 2.55    |
| 1,3-butadiene             | 1.75     | 2.15                             | 1.95    |
| Butane                    | 17.73    | 17.4                             | 17.57   |
| trans-2-butene            | 4.52     | 5.55                             | 5.03    |
| cis-2-butene              | 5.59     | 7.00                             | 6.29    |
| 3-methyl-1-butene         | 1.96     | 2.32                             | 2.14    |
| Isopentane                | 21.05    | 20.6                             | 20.83   |
| Pentene                   | 8.41     | 10.4                             | 9.41    |
| 2-methyl-2-butene         | 16.34    | 17.1                             | 16.72   |
| Pentane                   | 15.14    | 17.4                             | 16.27   |
| Isoprene                  | 0.80     | 1.21                             | 1.01    |
| trans-2-pentene           | 19.39    | 23.5                             | 21.45   |
| cis-2-pentene             | 9.88     | 11.7                             | 10.79   |
| 1,1-dimethyl cyclopropane | 28.35    | 27.5                             | 27.92   |
| 2,2-dimethyl butane       | 2.43     | 2.9                              | 2.66    |
| Cyclopentene              | 3.32     | 4.41                             | 3.87    |
| Cyclopentane              | 9.11     | 10.1                             | 9.61    |
| 2,3-dimethyl butane       | 6.25     | 6.22                             | 6.24    |
| 2-methyl pentane          | 27.80    | 26.4                             | 27.10   |
| Butanal *                 | 2.36     | 2.09                             | 2.23    |
| 3-methyl pentane          | 17.73    | 20.1                             | 18.92   |
| 2-methyl-1-pentene        | 0.90     | 2.37                             | 1.64    |
| Hexene                    | 0.14     | 0.33                             | 0.23    |
| Hexane                    | 20.41    | 20.7                             | 20.55   |
| 3-hexene                  | 1.60     | 1.56                             | 1.58    |
| 3-methyl-2-pentene        | 0.67     | 0.56                             | 0.62    |
| Methyl cyclopentane       | 10.43    | 10.4                             | 10.42   |
| 4-methyl-2-pentene        | 1.68     | 1.44                             | 1.56    |
| Benzene                   | 16.71    | 15.8                             | 16.26   |
| Cyclohexane               | 1.75     | 1.35                             | 1.55    |
| 2-methyl hexane           | 1.43     | 1.17                             | 1.30    |
| 2,3-dimethyl pentane      | 0.47     | 0.45                             | 0.46    |
| 3-methyl hexane           | 1.80     | 2.07                             | 1.94    |
| cis-1,3-dimethyl pentane  | 0.31     | 0.33                             | 0.32    |



Table 1: (Continued).

| VOC                          | Sample 1 | $\mu\text{g m}^{-3}$<br>Sample 2 | Average |
|------------------------------|----------|----------------------------------|---------|
| Heptane                      | 1.39     | 1.20                             | 1.29    |
| methyl cyclohexane           | 0.76     | 0.67                             | 0.71    |
| Toluene                      | 24.10    | 24.60                            | 24.35   |
| 2-methyl-heptane             | 0.26     | 0.33                             | 0.29    |
| 3-methyl-heptane             | 0.28     | 0.23                             | 0.25    |
| ethyl benzene                | 5.76     | 5.50                             | 5.63    |
| Benzaldehyde *               | 2.67     | 2.55                             | 2.61    |
| p-xylene                     | 11.17    | 10.70                            | 10.94   |
| m-xylene                     | 13.57    | 12.40                            | 12.99   |
| Styrene                      | 4.32     | 4.05                             | 4.19    |
| o-xylene                     | 10.71    | 10.40                            | 10.56   |
| Nonane                       | 0.62     | 0.78                             | 0.70    |
| isopropyl benzene            | 1.72     | 1.57                             | 1.64    |
| propyl benzene               | 1.07     | 1.08                             | 1.08    |
| 1-ethyl-4-methyl benzene     | 2.04     | 2.00                             | 2.02    |
| 1-ethyl-3-methyl benzene     | 1.45     | 1.44                             | 1.44    |
| 1,3,5 trimethyl benzene      | 1.55     | 1.87                             | 1.71    |
| 1-ethyl-2-methyl benzene     | 1.14     | 1.03                             | 1.08    |
| 1,2,4 trimethyl benzene      | 2.67     | 2.76                             | 2.71    |
| Decane                       | 0.76     | 0.94                             | 0.85    |
| 1,2,3-trimethyl benzene      | 1.27     | 1.12                             | 1.19    |
| Indane                       | 0.90     | 0.78                             | 0.84    |
| 1-methyl-3-propyl benzene    | 1.00     | 1.04                             | 1.02    |
| 4-ethyl 1,2 dimethyl benzene | 1.16     | 1.48                             | 1.32    |
| Naphthalene                  | 0.90     | 1.05                             | 0.97    |
| Undecane                     | 0.27     | 0.33                             | 0.30    |
| Total                        | 392      | 413                              |         |

\*VOC sampled using TO-11A methodology. All other sampled using TO-15 methodology.

The first compound to be adjusted is CO, as it is a primary pollutant with low reactivity. After performing the best possible fit, changing only the hourly CO emissions, a second adjustment was performed on the mixing layer height hourly values. It is important to remember that this adjustment is conducted around the average values obtained by the INEA emission inventory [19]. Because the mixed layer height directly influences the pollutants' concentrations, these values were adjusted to obtain the best possible fit between the simulated and measured CO concentration values. This value was taken as a basis for the mixing layer height and is the only measure recorded at 10:00 AM in Rio de Janeiro city.

After the simulated CO profile was adjusted against the experimental values, the next step was the adjustment of the NO<sub>x</sub> profile. This step was performed as an adjustment on hourly NO<sub>x</sub> emissions to reproduce the experimental values, similarly to that performed for CO, without changing the profile of the mixed layer set above. To achieve a better adjustment of CO and NO<sub>x</sub> profiles, a final adjustment in their deposition rates was also made between the acceptable limits (1.0 to 1.5 cm s<sup>-1</sup> for CO, 0.10 to 0.20 cm s<sup>-1</sup> for NO and 0.30 to 0.80 cm s<sup>-1</sup> to NO<sub>2</sub>) [20].

After adjusting the CO and NO<sub>x</sub> profiles, the next step was to adjust the simulated and experimental results for O<sub>3</sub>. As ozone is not emitted by any significant source, but rather formed by the reactions of VOC with NO<sub>x</sub> and

sunlight, the adjustment was made by considering the VOC hourly emissions, while emissions of NO<sub>x</sub> were adjusted. As the model does not provide data on hourly concentrations of VOC, the adjustment in VOC emissions is achieved by adjusting the O<sub>3</sub> profile.

The O<sub>3</sub> adjustment presented the greatest difficulties to fit simulated and experimental values. To be able to increase the O<sub>3</sub> level in the early hours, it was necessary to significantly increase the VOC and NO<sub>x</sub> emissions. However, changing these values contributed to a mismatch in the profile of NO<sub>x</sub> and an exaggerated increase in the O<sub>3</sub> peak. Moreover, it was not possible to adjust the O<sub>3</sub> profile after 1:00 PM. However, the O<sub>3</sub> profile was modelled in other studies that also used OZIPR coupled with SAPRC [6, 11] and showed a similar profile to others modelled in this study. Therefore, the lag of the O<sub>3</sub> maximum value was not considered relevant because the maximum O<sub>3</sub> model concentrations and experimental concentrations showed strong agreement. Results for O<sub>3</sub> are presented in Figure 2.

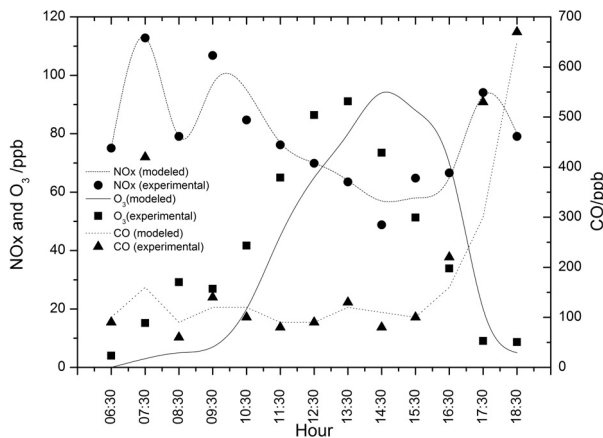


Figure 2: Results for experimental and modeled values of O<sub>3</sub>, CO and NO<sub>x</sub> for August, 29.

### 3.1 Vehicular emission

Due to difficulty in finding detailed information on the fleet of motorcycles in Brazil, a comparative study between the motorcycles emissions rate used in this work and the light vehicular fleet in use today was conducted.

To approach the real contributions of these vehicles regarding air quality, the survey was conducted in four stages. Initially, the light vehicles licensing inventory by company and model was consulted [21]. According to this inventory, approximately 77% of the fleet consists of vehicles manufactured by four companies.

The second step was to observe the amount of sales per model [22], which indicates that the best-selling car models in the country were seven models.

In the third step, the emission rates for the best-selling models and the average values were verified and consolidated.

Automobiles and motorcycles constitute approximately 88% of the fleet in Rio de Janeiro city. Thus, the estimated emissions rates of these two types of vehicles can be considered representative for the whole of the current fleet [21].

### 3.2 Scenario study

For the creation of the scenarios, it is necessary to estimate the expected emissions increase from motorcycles. Unfortunately, there is no database available for CO, NO<sub>x</sub> and HC emissions for motorcycles as there is for light vehicles. Thus, the average emissions values of two standard motorcycles were used, as described in the Methodology section for the dynamometer tests.

A comparison between average motorcycles and light vehicles emissions data is presented in Figure 3. It is important to remember that it is a conservative study, as the two evaluated motorcycles are new and attend the actual legislation.

Despite the difference in the methodologies used to measure the emissions (EC/97/24 for motorcycles and NBR 6601 for light vehicles), the average emissions found for the two motorcycles indicate higher concentrations for ethene, propene, ethyne, pentane, benzene and toluene. All of these pollutants presented average emission rates above 4 mg km<sup>-1</sup>.

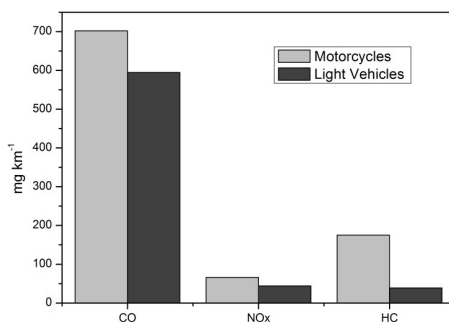


Figure 3: Comparison of emissions of motorcycles and light vehicles.

Regarding the fleet increase, which presents a quantitative history of motorcycles in use in Rio de Janeiro, a chart with a trend line fleet growth was prepared (Figure 4). From this graph, it was possible to obtain the estimated numbers of motorcycles for the next 3, 5 and 10 years starting from October 2011.

According to Figure 4, the motorcycle fleet will number approximately 300,000, representing an increase of 34% over the year 2011. Two years later, the increase will be almost 61%, to 360,000 motorcycles. Finally, in 10 years, the fleet growth will be approximately 132%, with approximately 520,000 motorcycles in circulation.

Based on these data, three scenarios with simulations to estimate the impact for the next 3, 5 and 10 years were created.



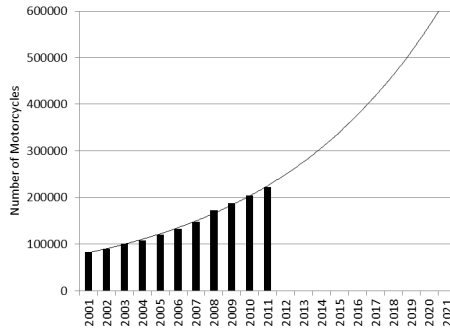


Figure 4: Trendline for the motorcycle fleet increases for the next 3, 5, and 10 years in Rio de Janeiro city.

To conduct scenarios studies, some assumptions are required. The atmosphere of a city is a combination of numerous parameters, such as emissions, topography, meteorology, and others. To choose a hypothetical scenario for a future condition of a city, it is not possible to assume changes in all parameters simultaneously because the errors associated may increase or even vanish. Thus, the study of scenarios here will take into account the following assumptions:

- Data for August 29 were used to fit the model (profiles of temperature, humidity, mixing layer and deposition rates);
- The VOC speciation of August 29 was used;
- The fleet increase of light or heavy vehicles was not considered;
- A possible variation in the composition of fuels used was not considered;
- Future reductions in emissions from vehicles and motorcycles that will surely be implemented in the future were not considered.

Thus, increases in CO, HC and NO<sub>x</sub>, arising from an increased motorcycle fleet, will be computed on the basis of their percentage in the fleet, leaving the hourly emissions used in the adjustment of the model weighted by increased emissions and the percentage of the fleet in MRRJ.

Assuming an increased motorcycles fleet, there will be a larger percentage of each scenario, as the vehicles have stabilized in number. The current percentage of the motorcycles in the MRRJ is 11.68%. Thus, the estimated percentage of motorcycles in three years is 15.65%, in 5 years is 18.81% and in 10 years is 27.10%.

It is also necessary to take into consideration that an increase in the motorcycle fleet will not lead to a linear increase in CO, NO<sub>x</sub> and HC, as motorcycles have higher emission rates for all pollutants, for instance, 13.44% higher than the vehicles for CO, 36.36% for NO<sub>x</sub> and 336% for HC.

The results for the ozone concentration profile in the troposphere for the scenarios studied are shown in Figure 5. It is possible to see that increasing the motorcycle fleet implies a significant increase in the ozone concentration in the troposphere. This increase is due to the high hydrocarbons emissions rate inherent in the operation of motorcycles compared to that of light vehicles.

Currently, the ozone concentration in the troposphere of Rio de Janeiro city is relatively low, as it is routinely below 100 ppb. This fact stems from the high concentration of NO<sub>x</sub> found.

According to the model results for the scenarios chosen, the highest ozone values were recorded around 2:30 PM, reaching 120 ppb for scenario 1, 130 ppb for scenario 2 and 160 ppb for scenario 3. Regarding air quality standards in accordance with those seen in Figure 5, one can predict that the reference values, national and international, will be exceeded in approximately three years. One exception to this estimate is the standard used by the EPA (320 ppb), which will not be reached.

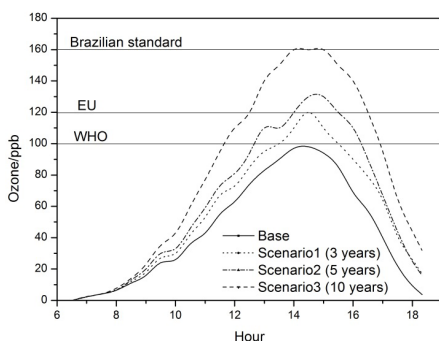


Figure 5: Ozone profile modeled for each scenario proposed.

Besides the studied scenarios results, the model was also used in multiple simulations for different maximum ozone concentrations of several VOC and NO<sub>x</sub> concentrations, providing data that generate an isopleth plot (lines of equal concentrations).

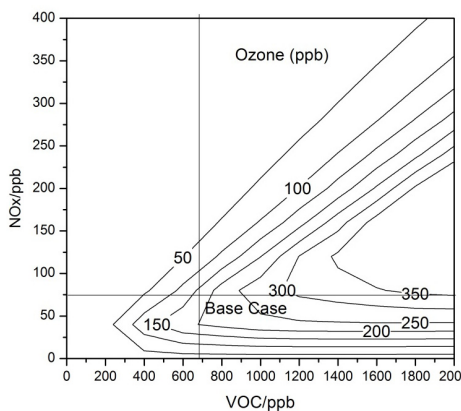


Figure 6: Ozone isopleth plot (ppb) for several concentrations of VOC and NO<sub>x</sub> for the base case on August, 29.

In Figure 6, isopleths for maximum ozone levels of the base case can be viewed. The figure shows a scenario with little variation in the NO<sub>x</sub> rate plus a large addition of VOC, resulting in significant increases in ozone.

## 4 Conclusions

From the results obtained in this work, a significant deterioration in air quality in large cities is expected, due to increases in motorcycles fleets in urban centres and the consequent increase in emissions. In 2001, the number of motorcycles (approximately 82,000) was less than 5% of the vehicular fleet of Rio de Janeiro city. In ten years, that number is expected to exceed a rate of 9% of the total fleet (approximately 224,000) and is growing steadily.

The results of VOC speciation emitted from motorcycles indicated high emission rates for very reactive species such as ethene, propene, ethyne, pentane, benzene and toluene. All of these pollutants presented average emission rates higher than 4 mg km<sup>-1</sup>.

The determined results indicate that the O<sub>3</sub> concentration in the troposphere will exceed the national standard established by the CONAMA Resolution 03/90 within three years.<sup>25</sup> It also indicates that the emission rates of motorcycles used in this study (HC, CO and NO<sub>x</sub>) already meet the levels recommended in all phases of PROMOT. Thus, in addition to the adoption of PROMOT, additional actions are necessary to manage the pollutants emission generated from mobile sources.

## Acknowledgements

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